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c. only carbon. d. only carbon and hydrogen.

2. The number of covalent bonds a carbon atom can form with other atoms is 3.

A covalent bond formed when two atoms share two pairs of electrons is called a

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a. -single bond. b.

double bond. 4. The breakdown of a polymer involves a. hydrolysis.

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Section 2-3 Carbon Compounds. (pages 44-48) This section explains how the element carbon is able to form millions of carbon, or organic, compounds. It also

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describes the four groups of organic compounds found in living things. The Chemistry of Carbon.

Macromolecules The Chemistry of Carbon

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VOCABULARY REVIEW

1. An organic compound is a compound containing carbon atoms covalently bonded to other carbon atoms and to other elements.

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Examples: any carbon-containing compound, such as benzene, ethanol, glycerol, glucose, fructose, sucrose, ATP, and ADP.

2.A functional group is a cluster of atoms in a com-

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