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### Solution Stoichiometry Chem Worksheet 15

8. If 525 mL of 0.80 M HCl solution is neutralized with 315 mL of Sr(OH) 2 solution what is the molarity of the Sr(OH) 2? \_\_\_ HCl + \_\_\_ Sr(OH) 2 \_\_\_ SrCl 2 + \_\_\_ H 2O Solution Stoichiometry Name \_\_\_\_\_ Chem Worksheet 15-6 xample How many grams of solid calcium hydroxide, Ca(OH) , are required to react with 350 mL of 0.40 M HCl?

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Solution Stoichiometry Worksheet Solve the following solutions Stoichiometry problems: 1. How many grams of silver chromate will precipitate when 150. mL of 0.500 M silver nitrate are added to 100. mL of 0.400 M potassium chromate? 2 AgNO 3(aq) + K 2 CrO 4(aq) Ag 2 CrO 4(s) + 2 KNO 3(aq) 0.150 L AgNO 3 0.500 moles AgNO 3 1 moles Ag 2 CrO 4 331 ...

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